



Edition: BP 2025 (Ph. Eur. 11.6 update)

Dexamfetamine Tablets

[General Notices](#)

Action and use

Amfetamine.

DEFINITION

Dexamfetamine Tablets contain Dexamfetamine Sulfate.

The tablets comply with the requirements stated under Tablets and with the following requirements.

Content of dexamfetamine sulfate, $(C_9H_{13}N)_2 \cdot H_2SO_4$

90.0 to 110.0% of the stated amount.

IDENTIFICATION

A. Dissolve a quantity of the powdered tablets containing 0.1 g of dexamfetamine sulfate as completely as possible in 20 mL of [water](#), filter, add 2 mL of 5M [sodium hydroxide](#) and extract with three 25 mL quantities of [ether](#), washing the combined extracts with 5 mL of [water](#). To the ether solution add 10 mL of 0.05M [sulfuric acid](#) and shake well. The acid layer, after warming to dispel residual ether and cooling to 20°, is dextrorotatory.

B. Extract a quantity of the powdered tablets containing 50 mg of dexamfetamine sulfate with 10 mL of [water](#), filter, cool to about 15°, add 3 mL of 1M [sodium hydroxide](#) and shake for 2 minutes with 1 mL of a mixture of 1 volume of [benzoyl chloride](#) and 2 volumes of [ether](#). Filter, wash the residue with 15 mL of [water](#) and recrystallise twice from [ethanol](#) (50%). The [melting point](#) of the crystals, after drying at 105° for 1 hour, is about 156°, [Appendix V A](#).

ASSAY

Weigh and powder 20 tablets, or more if necessary. Dissolve a quantity of the powder containing 0.1 g of Dexamfetamine Sulfate as completely as possible in 20 mL of [water](#), add 8 g of [sodium chloride](#) and 2 mL of 5M [sodium hydroxide](#) and extract with successive quantities of 50, 20, 20 and 20 mL of [ether](#). Extract the combined ether extracts with four 10 mL quantities of 0.1M [hydrochloric acid](#) and make the combined acid extracts alkaline with 5M [sodium hydroxide](#). Dilute to 120 mL with [water](#) and distil into 20 mL of [0.05M hydrochloric acid VS](#) until only 5 mL of liquid remains in the distillation flask. Boil, cool and titrate the excess of acid with [0.05M sodium hydroxide VS](#) using [methyl red solution](#) as indicator. Each mL of [0.05M hydrochloric acid VS](#) is equivalent to 9.212 mg of $(C_9H_{13}N)_2 \cdot H_2SO_4$.