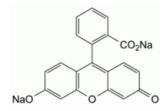
Edition: BP 2025 (Ph. Eur. 11.6 update)

Fluorescein Sodium

General Notices

Soluble Fluorescein

(Ph. Eur. monograph 1213)



C₂₀H₁₀Na₂O₅ 376.3 518-47-8

Action and use

Detection of corneal lesions, retinal angiography and pancreatic function testing.

Preparations

Fluorescein Eye Drops

Fluorescein Injection

Ph Eur

DEFINITION

Disodium 2-(6-oxido-3-oxo-3*H*-xanthen-9-yl)benzoate.

Content

95.0 per cent to 103.0 per cent (dried substance).

CHARACTERS

Appearance

Orange-red, fine powder, hygroscopic.

Solubility

Freely soluble in water, soluble in ethanol (96 per cent), practically insoluble in hexane and in methylene chloride.

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IDENTIFICATION

First identification: B, D.

Second identification: A, C, D.

A. Dilute 0.1 mL of solution S (see Tests) to 10 mL with <u>water R</u>. The solution shows yellowish-green fluorescence. The fluorescence disappears on addition of 0.1 mL of <u>dilute hydrochloric acid R</u> and reappears on addition of 0.2 mL of <u>dilute sodium hydroxide solution R</u>.

B. Infrared absorption spectrophotometry (2.2.24).

Preparation Discs.

Comparison Ph. Eur. reference spectrum of fluorescein sodium.

- C. The absorption by a piece of filter paper of 0.05 mL of the solution prepared for identification A (before the addition of <u>dilute hydrochloric acid R</u>) colours the paper yellow. On exposing the moist paper to bromine vapour for 1 min and then to ammonia vapour, the colour becomes deep pink.
- D. Ignite 0.1 g in a porcelain crucible. Dissolve the residue in 5 mL of <u>water R</u> and filter. 2 mL of the filtrate gives reaction (a) of sodium (2.3.1).

TESTS

Solution S

Dissolve 1.0 g in carbon dioxide-free water R prepared from distilled water R and dilute to 50 mL with the same solvent.

Appearance of solution

Solution S is clear (2.2.1) and orange-yellow with yellowish-green fluorescence.

pH (2.2.3)

7.0 to 9.0 for solution S.

Related substances

Liquid chromatography (2.2.29).

Test solution (a) Dissolve 0.100 g of the substance to be examined in a mixture of 30 volumes of <u>acetonitrile R</u> and 70 volumes of mobile phase A and dilute to 100.0 mL with the same mixture of solvents.

Test solution (b) Dilute 5.0 mL of test solution (a) to 250.0 mL with a mixture of 30 volumes of <u>acetonitrile R</u> and 70 volumes of mobile phase A.

Reference solution (a) Dissolve 55.0 mg of <u>diacetylfluorescein CRS</u> in a mixture of 1 mL of <u>2.5 M sodium hydroxide</u> and 5 mL of <u>ethanol (96 per cent) R</u>, heat on a water-bath for 20 min mixing frequently, cool and dilute to 50.0 mL with <u>water R</u>. Dilute 5.0 mL of the solution to 250.0 mL with a mixture of 30 volumes of <u>acetonitrile R</u> and 70 volumes of mobile phase A.

Reference solution (b) Dissolve 10.0 mg of phthalic acid R (impurity B) and 10.0 mg of resorcinol R (impurity A) in a mixture of 30 volumes of acetonitrile R and 70 volumes of mobile phase A and dilute to 100.0 mL with the same mixture of solvents. Dilute 5.0 mL of the solution to 100.0 mL with a mixture of 30 volumes of acetonitrile R and 70 volumes of mobile phase A.

Reference solution (c) Dilute 5.0 mL of test solution (b) to 20.0 mL with a mixture of 30 volumes of <u>acetonitrile R</u> and 70 volumes of mobile phase A.

Column:

— size: I = 0.25 m, $\emptyset = 4.6 \text{ mm}$;

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- stationary phase: <u>octylsilyl silica gel for chromatography R</u> (5 μm);
- temperature: 35 °C.

Mobile phase:

- *mobile phase A*: dissolve 0.610 g of *potassium dihydrogen phosphate R* in *water R* and dilute to 1000 mL with the same solvent; adjust to pH 2.0 with *phosphoric acid R*;
- mobile phase B: <u>acetonitrile for chromatography R</u>;

Time (min)	Mobile phase A (per cent <i>V/V</i>)	Mobile phase B (per cent <i>V/V</i>)
0 - 20	85 → 20	15 → 80
20 - 29	20	80
29 - 30	20 → 85	80 → 15
30 - 35	85	15

Flow rate 1.0 mL/min.

Detection Spectrophotometer at 220 nm.

Injection 20 µL of test solution (a) and reference solutions (b) and (c).

Relative retention With reference to fluorescein (retention time = about 15 min): impurity A = about 0.4; impurity B = about 0.5; impurity C = about 0.9.

System suitability Reference solution (b):

— <u>resolution</u>: minimum 1.5 between the peaks due to impurity A and impurity B.

Limits:

- correction factor: for the calculation of content, multiply the peak area of impurity C by 1.6;
- *impurities A, B*: for each impurity, not more than the area of the corresponding peak in the chromatogram obtained with reference solution (b) (0.5 per cent);
- *impurity C*: not more than the area of the principal peak in the chromatogram obtained with reference solution (c) (0.5 per cent);
- *unspecified impurities*: for each impurity, not more than 0.2 times the area of the principal peak in the chromatogram obtained with reference solution (c) (0.10 per cent);
- sum of impurities other than A, B, C: not more than the area of the principal peak in the chromatogram obtained with reference solution (c) (0.5 per cent);
- *disregard limit*: 0.1 times the area of the principal peak in the chromatogram obtained with reference solution (c) (0.05 per cent).

Chlorides (2.4.4)

Maximum 0.25 per cent.

To 10 mL of solution S add 90 mL of <u>water R</u> and 1 mL of <u>dilute nitric acid R</u>, wait for at least 10 min and filter. Dilute 10 mL of the filtrate to 15 mL with <u>water R</u>.

Sulfates (2.4.13)

Maximum 1.0 per cent.

To 5 mL of solution S add 90 mL of <u>distilled water R</u>, 2.5 mL of <u>dilute hydrochloric acid R</u> and dilute to 100 mL with <u>distilled water R</u>. Filter.

<u>Zinc</u>

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Dilute 5 mL of solution S to 10 mL with <u>water R</u>. Add 2 mL of <u>hydrochloric acid R1</u>, filter and add 0.1 mL of <u>potassium</u> <u>ferrocyanide solution R</u>. No turbidity or precipitate is formed immediately.

Loss on drying (2.2.32)

Maximum 10.0 per cent, determined on 1.000 g by drying in an oven at 105 °C.

ASSAY

Liquid chromatography (2.2.29) as described in the test for related substances with the following modification.

Injection Test solution (b) and reference solution (a).

Calculate the percentage content of $C_{20}H_{10}Na_2O_5$ using the chromatogram obtained with reference solution (a) and the declared content of <u>diacetylfluorescein CRS</u>.

1 mg of $\underline{\textit{diacetylfluorescein CRS}}$ is equivalent to 0.9037 mg of $C_{20}H_{10}Na_2O_5$.

STORAGE

In an airtight container, protected from light.

IMPURITIES

Specified impurities A, B, C.

A. benzene-1,3-diol (resorcinol),

B. benzene-1,2-dicarboxylic acid (phthalic acid),

C. 2-(2,4-dihydroxybenzoyl)benzoic acid.

Ph Eur