# **Quality standards**

Edition: BP 2025 (Ph. Eur. 11.6 update)

# **Ethylenediamine**

**General Notices** 

(Ph. Eur. monograph 0716)



C<sub>2</sub>H<sub>8</sub>N<sub>2</sub> 60.1 107-15-3

Ph Eur

### **DEFINITION**

Ethane-1,2-diamine.

### Content

98.0 per cent to 101.0 per cent.

### **CHARACTERS**

# **Appearance**

Clear, colourless or slightly yellow liquid, hygroscopic.

# Solubility

Miscible with water and with anhydrous ethanol.

On exposure to air, white fumes are evolved. On heating, it evaporates completely.

### **IDENTIFICATION**

- A. Relative density (2.2.5): 0.895 to 0.905.
- B. Boiling point (2.2.12): 116 °C to 118 °C.
- C. To 0.2 mL add 0.5 mL of <u>acetic anhydride R</u>. Boil. A crystalline mass forms after cooling, which dissolves in 5 mL of <u>propanol R</u> with heating. Cool the solution and add 5 mL of <u>ether R</u>. If necessary, initiate crystallisation by scratching the walls of the test-tube with a glass rod. Filter through a sintered-glass filter (<u>2.1.2</u>), wash with several portions of <u>ether R</u> and dry at 100-105 °C. The residue melts (<u>2.2.14</u>) at 173 °C to 177 °C.

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### **TESTS**

#### Solution S

Mix 10 g with carbon dioxide-free water R and dilute to 100 mL with the same solvent.

### Appearance of solution

Solution S is clear (2.2.1) and not more intensely coloured than the reference solution BY 6 (2.2.2, Method II).

### Carbonate

A mixture of 4 mL of solution S and 6 mL of <u>calcium hydroxide solution R</u> is not more opalescent than reference suspension II (<u>2.2.1</u>).

### **Chlorides** (2.4.4)

Maximum 100 ppm.

To 5 mL of solution S add 5 mL of dilute nitric acid R and dilute to 15 mL with water R.

#### Ammonia and other bases

Dissolve 1.2 g in 20 mL of <u>ethanol (96 per cent)</u> R and add, dropwise with stirring, 4.5 mL of <u>hydrochloric acid</u> R. Evaporate to dryness on a water-bath, breaking up any resulting cake with a glass rod, and dry at 100-105 °C for 1 h. 1 g of the residue is equivalent to 0.4518 g of  $C_2H_8N_2$ . Calculate the percentage content of  $C_2H_8N_2$ : it does not vary by more than 0.5 from the percentage content determined in the assay.

# Iron (2.4.9)

Maximum 10 ppm, determined on solution S.

# Residue on evaporation

Maximum 0.3 per cent.

Evaporate 5.00 g to dryness on a water-bath and dry at 100-105 °C for 1 h. The residue weighs a maximum of 15 mg.

### **ASSAY**

Place 25.0 mL of <u>1 M hydrochloric acid</u> and 0.2 mL of <u>methyl red mixed solution R</u> in a flask. Add 0.600 g of the substance to be examined. Titrate with <u>1 M sodium hydroxide</u> until the colour changes from violet-red to green.

1 mL of 1 M hydrochloric acid is equivalent to 30.05 mg of C<sub>2</sub>H<sub>8</sub>N<sub>2</sub>.

# **STORAGE**

In an airtight container, protected from light.

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