



Edition: BP 2025 (Ph. Eur. 11.6 update)

## Co-magaldrox Oral Suspension

### [General Notices](#)

Magnesium Hydroxide and [Aluminium Hydroxide Oral Suspension](#)

### Action and use

Antacid.

### DEFINITION

Co-magaldrox Oral Suspension is a suspension containing Magnesium Hydroxide and Dried Aluminium Hydroxide in a suitable flavoured vehicle. The amount of dried aluminium hydroxide is adjusted to give the required content of  $\text{Al}_2\text{O}_3$ .

*The oral suspension complies with the requirements stated under Oral Liquids and with the following requirements.*

### Content of magnesium hydroxide, $\text{Mg}(\text{OH})_2$

90.0 to 110.0% of the stated amount.

### Content of $\text{Al}_2\text{O}_3$

45.4 to 58.8% of the stated amount of dried aluminium hydroxide.

### IDENTIFICATION

- A. Mix 5 mL with 10 mL of [2M hydrochloric acid](#), add 5 drops of [methyl red solution](#) and heat to boiling. Add 6M [ammonia](#) until the solution becomes yellow, continue boiling for 2 minutes and filter. To 1 mL of the filtrate add 1 mL of 6M [ammonia](#) and 1 mL of 2M [ammonium chloride](#); no precipitate is produced. Add 0.25M [disodium hydrogen orthophosphate](#); a white, crystalline precipitate is produced which is insoluble in 6M [ammonia](#).
- B. To 5 mL add 10 mL of [2M hydrochloric acid](#). The solution yields the reaction characteristic [of aluminium salts](#), [Appendix VI](#).

### TESTS

#### Microbial contamination

Carry out a quantitative evaluation for enterobacteria and certain other Gram-negative bacteria, Appendix XVI B1. 0.01 mL of the preparation gives a negative result, Table I (most probable number of bacteria per gram fewer than  $10^2$ ).

### ASSAY

**For  $Al_2O_3$**

To a weighed quantity containing 1.5 g of dried aluminium hydroxide add 20 mL of [water](#), stir and slowly add 10 mL of [hydrochloric acid](#). Heat gently, if necessary, to aid solution, cool, filter, wash the filter well with [water](#), dilute the combined filtrate and washings to 200 mL with [water](#) and mix. Reserve a portion of the solution for the Assay for magnesium hydroxide. To 10 mL add 20 mL of [water](#) and, with continuous stirring, 25 mL of 0.05M [disodium edetate VS](#) followed by 20 mL of a mixture of equal volumes of 2M [ammonium acetate](#) and 2M [acetic acid](#). Heat near the [boiling point](#) for 5 minutes, cool, add 50 mL of [absolute ethanol](#) and 3 mL of a freshly prepared 0.025% w/v solution of [dithizone](#) in [absolute ethanol](#). Titrate the excess of disodium edetate with [0.05M zinc sulfate VS](#) until the colour of the solution changes from greenish blue to reddish violet. Each mL of 0.05M [disodium edetate VS](#) is equivalent to 2.549 mg of  $Al_2O_3$ . Determine the [weight per mL](#) of the suspension, [Appendix V G](#), and calculate the percentage content of  $Al_2O_3$ , weight in volume.

**For [magnesium hydroxide](#)**

To a volume of the solution reserved in the Assay for  $Al_2O_3$  containing about 40 mg of magnesium hydroxide add 200 mL of [water](#) and 20 mL of [triethanolamine](#) and stir. Add 10 mL of [ammonia buffer pH 10.9](#) and cool the solution to between 3° and 4° by immersion in iced water. Titrate the cooled solution with 0.05M [disodium edetate VS](#) using [mordant black 11 solution](#) as indicator. Each mL of 0.05M [disodium edetate VS](#) is equivalent to 2.916 mg of  $Mg(OH)_2$ . Using the [weight per mL](#) of the suspension, calculate the percentage content of  $Mg(OH)_2$ , weight in volume.

**STORAGE**

Co-magaldrox Oral Suspension should not be allowed to freeze.