Edition: BP 2025 (Ph. Eur. 11.6 update)

Apramycin Sulfate

General Notices

Apramycin Sulphate

HO OH OH NHMe OH
$$R = H$$

$$R =$$

C₂₁H₄₁N₅O₁₁,2½H₂SO₄ 784.8 41194-16-5

Action and use

Aminoglycoside antibacterial.

Preparations

Apramycin Veterinary Oral Powder

Apramycin Premix

DEFINITION

Apramycin Sulfate is the sulfate of 4-*O*-[(2*R*,3*R*,4a*S*,6*R*,7*S*,8*R*,8a*R*)-3-amino-6-(4-amino-4-deoxy-α-D-glucopyranosyloxy)-8-hydroxy-7-methylaminoperhydropyrano[3,2-*b*]pyran-2-yl]-2-deoxystreptamine. It is produced by the growth of certain strains of *Streptomyces tenebrarius* or obtained by any other means. The potency is not less than 450 Units per mg, calculated with reference to the anhydrous substance.

CHARACTERISTICS

A light brown powder or granular material; hygroscopic.

Freely soluble in water, practically insoluble in acetone, in ethanol (96%), in ether and in methanol.

IDENTIFICATION

- A. Carry out the method for thin-layer chromatography, Appendix III A, using the following solutions in water.
- (1) 0.1% w/v of the substance being examined.
- (2) 0.07% w/v of apramycin BPCRS.
- (3) 0.07% w/v of each of apramycin BPCRS and tobramycin BPCRS.

CHROMATOGRAPHIC CONDITIONS

- (a) Use a silica gel F₂₅₄ precoated plate (Merck <u>silica gel 60 F₂₅₄ plates are suitable</u>).
- (b) Use the mobile phase as described below.
- (c) Apply 5 µL of each solution.
- (d) Develop the plate to 10 cm.
- (e) After removal of the plate, allow it to dry in a current of warm air, spray with a mixture of equal volumes of a 46% w/v solution of <u>sulfuric acid</u> and a 0.2% w/v solution of <u>naphthalene-1,3-diol</u> in <u>ethanol (96%)</u> and heat at 150° for 5 to 10 minutes.

MOBILE PHASE

20 volumes of <u>chloroform</u>, 40 volumes of 13.5M <u>ammonia</u> and 60 volumes of <u>methanol</u>, equilibrated for 1 hour before use.

SYSTEM SUITABILITY

The test is not valid unless the chromatogram obtained with solution (3) shows two clearly separated principal spots.

CONFIRMATION

The principal spot in the chromatogram obtained with solution (1) corresponds in colour and position to that in the chromatogram obtained with solution (2).

- B. In the test for Related substances, the retention time of the principal peak in the chromatogram obtained with solution (1) corresponds to that of the principal peak in the chromatogram obtained with solution (2).
- C. Yields reaction A characteristic of sulfates, Appendix VI.

TESTS

Sulfate

26.0 to 33.0% of SO₄, calculated with reference to the anhydrous substance, when determined by the following method. Dissolve 0.25 g in 100 mL of <u>water</u>, adjust to pH 11 with 13.5м <u>ammonia</u> and add 10 mL of <u>0.1м barium chloride VS</u>. Titrate with <u>0.1м disodium edetate VS</u> using 0.5 mg of <u>phthalein purple</u> as indicator; add 50 mL of <u>ethanol (96%)</u> when the colour of the solution begins to change and continue the titration until the violet-blue colour disappears. Each mL of <u>0.1м barium chloride VS</u> is equivalent to 9.606 mg of SO₄.

Caerulomycin and dipyridyl derivatives

Dissolve 0.5 g in <u>water</u> in a 100 mL graduated flask, add 10 mL of <u>methanol</u> and dilute to 100 mL with <u>water</u>. Place 5 mL in a 25 mL graduated flask and add 5 mL of an acetate buffer prepared by dissolving 8.3 g of <u>anhydrous sodium acetate</u> in 25 mL of <u>water</u>, adding 12 mL of <u>glacial acetic acid</u> and diluting to 100 mL with <u>water</u>. Mix, add 1 mL of a 10% w/v solution of <u>hydroxylamine hydrochloride</u> in <u>water</u> and mix again. Add 5 mL of a 1% w/v solution of <u>ammonium iron(II) sulfate</u> and dilute to 25 mL with <u>water</u>. Measure the <u>absorbance</u> of the resulting solution at 520 nm, <u>Appendix II B</u>, using in the reference cell a solution obtained by carrying out the same procedure without the substance being examined. The <u>absorbance</u> is not greater than that obtained by repeating the test using 5 mg of 2,2'-dipyridyl dissolved in 10 mL of <u>methanol</u> and diluted to 100 mL with <u>water</u> and beginning at the words 'Place 5 mL in a 25 mL graduated flask...' (1%).

Related substances

Carry out the method for liquid chromatography, Appendix III D, using the following solutions in water.

- (1) 0.50% w/v of the substance being examined.
- (2) 0.35% w/v of <u>apramycin BPCRS</u>.
- (3) Dilute 1 volume of solution (1) to 20 volumes.
- (4) Dilute 1 volume of solution (3) to 50 volumes.

CHROMATOGRAPHIC CONDITIONS

- (a) Use a column (25 cm \times 4 mm) packed with fast cation-exchange polymeric beads (13 μ m) with sulfonic acid functional groups (Dionex Fast Cation-1^R is suitable) and a stainless steel post-column reaction coil (380 cm \times 0.4 mm) with internal baffles. Use in the reaction coil *ninhydrin reagent I* at a flow rate approximately the same as that for the mobile phase.
- (b) Use gradient elution and the mobile phase described below.
- (c) Use a flow rate of 0.8 mL per minute.
- (d) Use a column temperature of 130°. Maintain the post-column reaction coil at the same temperature.
- (e) Use a detection wavelength of 568 nm.
- (f) Inject 20 μL of each solution.

MOBILE PHASE

Mobile phase A A solution containing 1.961% w/v of <u>sodium citrate</u>, 0.08% v/v of <u>liquefied phenol</u> and 0.5% v/v of <u>thiodiglycol</u>, adjusted to pH 4.25 using <u>hydrochloric acid</u>.

Mobile phase B A solution containing 4.09% w/v of <u>sodium chloride</u> and 3.922% w/v of <u>sodium citrate</u> with 0.08% v/v of <u>liquefied phenol</u>, adjusted to pH 7.4 with <u>hydrochloric acid</u>.

Equilibrate the column using a mixture containing 75% of mobile phase A and 25% of mobile phase B. After each injection elute for 3 minutes using the same mixture and then carry out a linear gradient elution for 6 minutes to 100% of mobile phase B. Elute for a further 21 minutes using 100% of mobile phase B, then step-wise re-equilibrate to a mixture of 75% of mobile phase A and 25% of mobile phase B and elute for at least 10 minutes.

SYSTEM SUITABILITY

The test is not valid unless, in the chromatogram obtained with solution (2), the <u>resolution factor</u> between the peaks due to compound A and 3-hydroxyapramycin, identified using the reference chromatogram supplied with <u>apramycin BPCRS</u>, is at least 0.8.

LIMITS

Multiply the areas of all the <u>secondary peaks</u> by 0.5. [NOTE: This is to ensure that the impurities are calculated relative to Apramycin Sulfate (which contains about 50% w/w of apramycin).]

In the chromatogram obtained with solution (1):

the areas of any peaks corresponding to 3-hydroxyapramycin, lividamine/2-deoxystreptamine (combined), compound A and compound B (identified using the reference chromatogram supplied with <u>apramycin BPCRS</u>) are not greater than 1.4, 1.0, 0.4 and 0.4 times respectively the area of the principal peak in the chromatogram obtained with solution (3) (7%, 5%, 2% and 2% respectively);

the area of any other <u>secondary peak</u> is not greater than 0.4 times the area of the principal peak in the chromatogram obtained with solution (3) (2%);

the sum of the areas of all the <u>secondary peaks</u> is not greater than 3 times the area of the principal peak in the chromatogram obtained with solution (3) (15%).

Disregard any peak with an area less than the area of the peak in the chromatogram obtained with solution (4) (0.1%).

Sulfated ash

Not more than 1.0%, Appendix IX A, Method II. Use 1 g.

Water

Not more than 14.0% w/w, <u>Appendix IX C</u>. Use 0.2 g and 20 mL of a mixture containing 1 volume of <u>methanol</u> and 2 volumes of <u>formamide</u> as the solvent. The solvent mixture must be prepared at least 12 hours before use and should be stored in an airtight container.

ASSAY

Carry out the <u>microbiological assay of antibiotics</u>, <u>Appendix XIV A</u>, Method B. The precision of the assay is such that the fiducial limits of error are not less than 95% and not more than 105% of the estimated potency.

IMPURITIES

A. caerulomycin,

$$HO$$
 O
 OH
 NH_2
 OH
 NH_2

B. lividamine,

$$\begin{array}{c} \text{HO} \\ \text{HO} \\ \\ \text{NH}_2 \end{array}$$

- C. 2-deoxystreptamine,
- D. 3-hydroxyapramycin; R = OH,
- E. 'compound A',
- F. 'compound B'.