## **Quality standards**

Edition: BP 2025 (Ph. Eur. 11.6 update)

## **Calcium Polystyrene Sulfonate**

**General Notices** 

Calcium Polystyrene Sulphonate

#### Action and use

Used in the treatment of hyperkalaemia.

## **DEFINITION**

Calcium Polystyrene Sulfonate is a cation-exchange resin prepared in the calcium form containing not less than 6.5% w/w and not more than 9.5% w/w of calcium, calculated with reference to the dried substance. Each g exchanges not less than 1.3 mEq and not more than 2.0 mEq of potassium, calculated with reference to the dried substance.

#### **CHARACTERISTICS**

A cream to light brown, fine powder.

Practically insoluble in water and in ethanol (96%).

#### **IDENTIFICATION**

- A. The <u>infrared absorption spectrum</u>, <u>Appendix II A</u>, is concordant with the <u>reference spectrum</u> of calcium polystyrene sulfonate <u>(RS 037)</u>.
- B. Yields reaction C characteristic of calcium salts, Appendix VI.

## **TESTS**

#### Particle size

Not more than 1% w/w is retained on a 150-µm sieve, Appendix XVII B. Use 20 g and sieve for 5 minutes.

### **Potassium**

Not more than 0.1% of K when determined by *atomic emission spectrophotometry*, <u>Appendix II D</u>, measuring at 766.5 nm and using a solution prepared in the following manner. To 1.1 g of the substance being examined add 5 mL of <u>hydrochloric acid</u>, heat to boiling, cool and add 10 mL of <u>water</u>. Filter, wash the filter and residue with <u>water</u> and dilute the filtrate and washings to 25 mL with <u>water</u>. Use <u>potassium standard solution (100 ppm K)</u>, suitably diluted with <u>water</u>, to prepare the <u>standard solutions</u>.

#### Sodium

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Not more than 0.1% of Na when determined by *atomic emission spectrophotometry*, <u>Appendix II D</u>, measuring at 589.0 nm and using a solution prepared in the following manner. To 1.1 g of the substance being examined add 5 mL of <u>hydrochloric acid</u>, heat to boiling, cool and add 10 mL of <u>water</u>. Filter, wash the filter and residue with <u>water</u> and dilute the filtrate and washings to 25 mL with <u>water</u>. Use <u>sodium solution</u> (200 ppm Na), suitably diluted with <u>water</u>, to prepare the <u>standard solutions</u>.

#### **Arsenic**

1 g dispersed in 25 mL of water complies with the limit test for arsenic, Appendix VII (1 ppm).

#### Styrene

Carry out the method for <u>liquid chromatography</u>, <u>Appendix III D</u>, using the following solutions.

- (1) Shake 10 g of the substance being examined with 10 mL of <u>acetone</u> for 30 minutes, centrifuge and use the supernatant liquid.
- (2) 0.0001% w/v of styrene in acetone.

#### CHROMATOGRAPHIC CONDITIONS

- (a) Use a stainless steel column (30 cm × 4 mm) packed with <u>octadecylsilyl silica gel for chromatography</u> (μBondapak C18 is suitable).
- (b) Use isocratic elution and the mobile phase described below.
- (c) Use a flow rate of 2 mL per minute.
- (d) Use an ambient column temperature.
- (e) Use a detection wavelength of 254 nm.
- (f) Inject 20 µL of each solution.

MOBILE PHASE

Equal volumes of acetonitrile and water.

LIMITS

In the chromatogram obtained with solution (1):

the area of any peak corresponding to styrene is not greater than the area of the peak in the chromatogram obtained with solution (2) (1 ppm).

## Potassium exchange capacity

To 3 g of the substance being examined in a dry 250 mL glass-stoppered flask add 100 mL of a solution containing 0.7455% w/v of potassium chloride and 0.4401% w/v of potassium hydrogen carbonate in water (solution A), stopper and shake for 15 minutes. Filter and dilute 2 mL of the filtrate to 1000 mL with water. Determine the concentration of unbound potassium in this solution by atomic emission spectrophotometry, Appendix II D, measuring at 766.5 nm and using solution A suitably diluted with water, to prepare the standard solutions. Calculate the potassium exchange capacity of the substance being examined in milliequivalents taking the concentration of potassium in solution A as 144 milliequivalents of K per litre.

## Loss on drying

When dried at 70° at a pressure not exceeding 0.7 kPa for 16 hours, loses not more than 8.0% of its weight. Use 2 g.

#### Microbial contamination

Carry out a quantitative evaluation for Enterobacteria and certain other Gram-negative bacteria, Appendix XVI B1. 0.01 g of the substance being examined gives a negative result, Table I (most probable number of bacteria per gram fewer than 10<sup>2</sup>).

## **ASSAY**

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#### For calcium

Carefully heat 1 g in a platinum crucible until a white ash is obtained and dissolve in 10 mL of <u>2M hydrochloric acid</u> with the aid of heat. Transfer the resulting solution to a conical flask using 20 mL of <u>water</u>. Add 50 mL of <u>0.05M disodium edetate</u> <u>VS</u>, 20 mL of <u>ammonia buffer pH 10.9</u> and titrate the excess of disodium edetate with <u>0.02M zinc sulfate VS</u>, using a 0.5% w/v solution of <u>mordant black 11</u> in <u>ethanol (96%)</u> as indicator to a red purple end point. Each mL of <u>0.05M disodium</u> <u>edetate VS</u> is equivalent to 2.004 mg of Ca.

## **STORAGE**

Calcium Polystyrene Sulfonate should be kept in an airtight container.